

LETTERS TO THE EDITOR

Reaction of 4-Oxo-5,6-benzo-1,3,2-dioxaphosphinin-2-yl Isocyanate with Dialkyl Arylcarbonylphosphonates

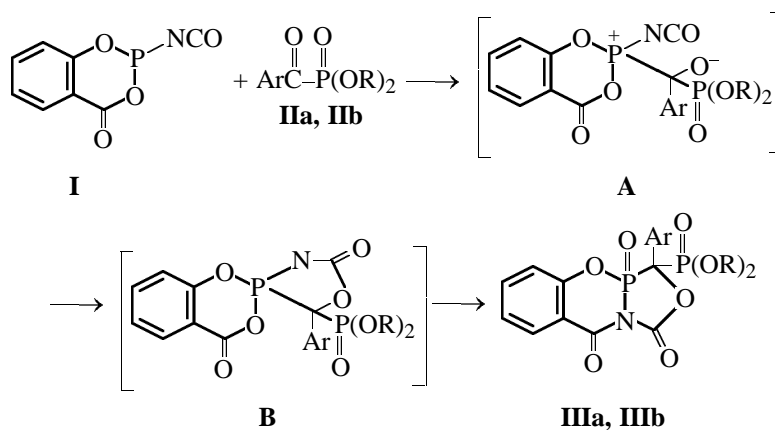
L. M. Burnaeva, V. F. Mironov, N. M. Azanchev, I. V. Konovalova,
G. A. Ivkova, O. V. Yashagina, and A. N. Pudovik

Kazan State University, Kazan, Tatarstan, Russia
Arbuzov Institute of Organic and Physical Chemistry, Kazan Research Center,
Russian Academy of Sciences, Kazan, Tatarstan, Russia

Received June 7, 2001

4-Oxo-5,6-benzo-1,3,2-dioxaphosphinin-2-yl isocyanate (**I**) (“salicyl isocyanatophosphite”), which contains both exo- and endocyclic carbonyl groups, can enter reactions with carbonyl compounds, involving the isocyanate group and yielding spiroposphoranes and diazadiphosphetidines, (2) involving the endocyclic carbonyl group and yielding ring-expansion products (6,7-benzo-1,3,2-dioxaphosphepines) [1], or (3) involving both functionalities and yielding phosphabicyclononanes with phosphorus–oxygen and phosphorus–nitrogen bonds [2].

Here we report for the first time that phosphite **I** can react with dialkyl arylcarbonylphosphonates **II** (80°C, 8 h, C₆H₆) to give cyclic compounds, viz. substituted phosphabicyclononanes **III** with a phosphorus–carbon bond. The cyclization process involving the isocyanate group probably occurs at the stage of formation of dipolar ion **A** by attack of the phosphorus atom of phosphite **I** on the electropilic center of phosphonate **II** (carbonyl carbon atom). Intermediate phosphimine **B** further undergoes the iminophosphonate–amidophosphonate rearrangement to give final product **III**.



Ar = *n*-ClC₆H₄ (**a**), 1-naphthyl (**b**); R = Et (**a**), Me (**b**).

The structures of the substituted phosphabicyclononanes were proved by physicochemical methods. The ³¹P–{¹H} NMR spectra contain two doublets at δ_p 8–10 and 11–13 ppm, characteristic of phosphonate phosphorus atoms. The stereoselectivity of the process

is quite high (diastereomeric ratio 9:1). The ¹³C NMR spectra contain double sets of signals of carbon atoms of the same kind, implying formation of two diastereomers, while the presence of a doubled doublet at 82.15–82.18 ppm with expected coupling constants

($^1J_{\text{PC}}$ 153.0–153.7 and 110.3–110.9 Hz) provide unequivocal evidence for the formation of a P–S–P fragment.

7-(*p*-Chlorophenyl)-2-(diethoxyphosphinoyl)-1-aza-3,4-benzo-5,8-dioxa-6-phosphabicyclo[4.2.0]-nonane-2,6,9-trione (IIIa). A mixture of 4.26 g of phosphite **I**, 5.64 g of diethyl (*p*-chlorophenylcarbonyl)phosphonate (**IIa**), and 10 ml of benzene was heated at 80°C for 8 h under dry argon. The reaction mixture was reprecipitated into pentane to isolate 6.93 g (70%) of compound **IIIa**, mp 164–166°C. IR spectrum, ν , cm^{-1} : 2965, 1835, 1720, 1660, 1605, 1490, 1455, 1370, 1310, 1285, 1255, 1205, 1170, 1115, 1090, 1040, 1010, 980, 940, 865, 820, 755, 720. ^{31}P – $\{^1\text{H}\}$ NMR spectrum (CH_2Cl_2), δ_{P} , ppm (J , Hz): 12.87 and 8.88 d.d ($^2J_{\text{PCP}}$ 12.7), d_2 , 22%; 11.40 and 9.72 d.d ($^2J_{\text{PCP}}$ 20.4), d_1 , 78%. Found, %: C 46.80; H 3.60; P 12.50. $\text{C}_{19}\text{H}_{18}\text{ClNO}_8\text{P}_2$. Calculated, %: C 46.96; H 3.70; P 12.77.

7-(Dimethoxyphosphinoyl)-7-naphthyl-1-aza-3,4-benzo-5,8-dioxa-6-phosphabicyclo[4.2.0]-nonane-2,6,9-trione (IIIb) was obtained similarly to compound **IIIa** from 4.93 g of phosphite **I** and 6.23 g of dimethyl (naphthylcarbonyl)phosphonate (**IIb**). Yield 6.74 g (60%), mp 174–176°C. IR spectrum, ν , cm^{-1} : 2960, 2930, 1835, 1720, 1675, 1612, 1515, 1460, 1400, 1380, 1345, 1315, 1300, 1225, 1185, 1160, 1120, 1090, 1065, 1030, 970, 950, 898, 880,

775, 760. ^{31}P – $\{^1\text{H}\}$ NMR spectrum ($\text{DMSO}-d_6$), δ_{P} , ppm (J , Hz): 9.42 and 17.0 d.d ($^2J_{\text{PCP}}$ 22.8). Found, %: C 53.33; H 3.71; P 12.92. $\text{C}_{21}\text{H}_{17}\text{NO}_8\text{P}_2$. Calculated, %: C 53.28; H 3.59; P 13.10.

The IR spectra were obtained on a Specord M-80 spectrometer in Vaseline oil. The NMR spectra were measured on Varian Unity-300 (^{31}P – $\{^1\text{H}\}$, 121.42 MHz) and Bruker MSL-400 (^{13}C , ^{13}C – $\{^1\text{H}\}$, 100.6 MHz) spectrometer against internal HMDS and external H_3PO_4 .

ACKNOWLEDGMENTS

The work was supported by the *Universitety Rossii* Program (project no. 015-05-01-17), the Program for Support of Leading Scientific Schools of the Russian Federation (project no. 00-15-97424), and the *Fundamental'nye issledovaniya i vysshee obrazovanie* Joint Program of CRDF and the Ministry of Education of the Russian Federation (REC-007).

REFERENCES

1. Konovalova, I.V., Burnaeva, L.M., Mironov, V.F., Khlopushina, G.A., and Pudovik, A.N., *Zh. Obshch. Khim.*, 1994, vol. 64, no. 1, p. 63.
2. Burnaeva, L.M., Mironov, V.F., Romanov, S.V., and Konovalova, I.V., *Zh. Obshch. Khim.*, 1999, vol. 69, no. 12, p. 2054.